Date:	June 9, 1993	
Job No.	7200-011	



Transmittal To:

Name I	Bill Dunn		
Firm	<u>City</u> of <u>Fresno,</u> Water Division		
Address	1910 E. University Ave.	<u> </u>	
City/State	Fresno, CA 93703-2988		

Transmittal From:

Name	Dave Murbach		<u>Department</u>	141	
Autoatte	ndant Number_	(916) 939-4100	Ext.	261	

Bill:

Attached you will find:

- 1) two sample kits for Volunteer Nos. 21 and 63
- 2) extra copy of residential sample collection procedures
- 3) example sample activity form

Before distributing the sample kits, fill in the day and date you want to pickup the samples (at the top of the instruction sheet). After you drop off the sample kit to the address shown in the lower right-hand corner of the instructions, you will need to call the volunteer (if you don't see them at the door) to let them know that they need to sample the next morning and see if they have any questions. When you pickup the sample bottle you need to confirm that the number on the bottle matches the number on the instruction sheet label, confirm that at least six hours passed between the last time water was used and the time the sample was taken, and complete the label on the sample bottle with the date and time the sample was taken as well as the last name of the sampler. Before delivery of the sample bottle to the lab (BSK at 1414 Stanislaus St.), you need to fill out a sample activity form as shown on the example I sent.

The residential instruction form is generic and can be used at any time by filling out the date at the top of the page and placing an address label with the volunteer number in the lower right-hand corner. You may need to use this if there are any other homes that need to schedule their sampling before 6/21.

Please call me if you have any questions. Thanks.

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Sample Detail

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Comments or Instructions:

PLEAS	SE TAKE THIS WATER SAMPLE FOR CITY OF FRESNO'S WATER DIVISION TOMORROW NING ON, _/_/93.
IF YO	U HAVE ANY QUESTIONS OR CANNOT TAKE THE SAMPLE FOR ANY REASON PLEASE CALL BILL IS AT 498-4136.
THAN	K YOU.
	<u>DIRECTIONS</u> <u>RESIDENT TAP SAMPLE COLLECTION PROCEDURES</u>
These solder Agend	e samples are being collected to determine the contribution of faucet fixtures and household pipes and/or to the lead and copper levels in tap water. This sampling effort is required by the Environmental Protection cy, and is being accomplished through the cooperation of homeowners and residents.
The c	collection procedure is described in detail below.
1.	Do not use any water for 6-8 hours before sampling. The water department recommends that either early mornings or evenings upon returning home from work are the best sampling times to ensure that the necessary water conditions exist.
2.	A kitchen or bathroom cold-water faucet is to be used for sampling. Place the open sample bottle below the faucet and gently open the cold water tap. Slowly fill the sample bottle to the base of the neck and turn off the water. It should take about 45 seconds to fill the bottle.
3.	Tightly cap the sample bottle and place in the plastic bag provided. Fill in the information requested on the form below, sign the form, and place in the plastic bag along with the sample bottle. Please review the address label below at this time to ensure that all information contained on the label is correct.
4.	Place the sample outside your home for pick-up by 9:00 in the morning.
5.	Results from this monitoring effort will be provided to participating customers when reports are generated for the State.
Plea	se call Bill Dunn at 498-4136 if you have any questions regarding these instructions.
TO	BE COMPLETED BY RESIDENT:
	EASE RETURN THIS SHEET WITH THE SAMPLE BOTTLE.
	ter was last used: TIME: DATE:
San	nple was collected: TIME: DATE:
I ha	ave read the above directions and have taken a tap sample in accordance with these directions.
	nature DATE:
	me:
	BE COMPLETED BY THE WATER DEPARTMENT EMPLOYEE:
Sar	mple picked up by: DATE:
0720	(011.008



1414 Stanislaus Street Fresno, California 93706 Telephone (209) 485-8310 FAX (209) 485-6935 1-800-877-8310

JUL 2 7 1993

Fresno City Water Division

Attn: Doug Kirk

1910 East University Ave.

Fresno, CA 93703

Date Sampled :

: 06/17/93

Time Sampled

: 0740

Date Received : 06/17/93

Report Issue Date: 06/30/93

Sample Type: LIQUID

Case Number

: Ch931597

Lab ID Number

: 1597

Project Number

: None

Sample Description: L & C-1

Results of Analyses EPA Lead and Copper Rule*

Compound	Method	Units	Results	DLR
Copper (Cu)	EPA 200.7	mg/L	0.025	0.010
Lead (Pb)	EPA 239.2	mg/L	0.0025	0.001
Turbidity	EPA 180.1	NTU	ND	0.1

KEY to ABBREVIATIONS

DLR: Detection Limit for the purposes of Reporting

mg/L: Milligrams per Liter

NTU: Nephelometric Turbidity Units

* The BPA Lead and Copper Rule includes several unique features. The sample is a "first draw" of a fixed volume (one liter). The turbidity analysis is performed after the sample is acidified with HNO3. Reporting near the detection limit is as follows:

Copper: Values less than 0.01 mg/L are reported as 0 (zero); Values between 0.010 and 0.049 mg/L are reported as 0.025 mg/L; Values greater than 0.05 mg/L are reported directly.

Lead: Values less than 0.001 mg/L are reported as 0 (zero); Values between 0.0010 and 0.0049 mg/L are reported as 0.0025 mg/L; Values greater than 0.005 mg/L are reported directly.

cc: HDR Engineering

Cynthia Pigman, QA/QC Supervisor

Doug

V. Inormanics Manager

8920916 LCR.T

Need copies when complete to HDR: Chain of Custodies (BSK) Field Logs Resident's Instructions (completed) Revise Distribution Address List Somple Log Sheet

Received by:	414	354	0	Date: 1/2/2 Time: 1: 2 am pm Lab Reference: No of Samples:
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Comments or Instructions:	ions:			

7-15-93

- Juvary Date: 712193 Time: 9: colum pm	Time:	75/ No of Samples:
Relinquished by: Allen	Date: 7/2/18/18	Lab Reference:
Sampled by: ARTHUR MANNERO	Received by: ADMADA (DOCOC)	Laboratory: 1984, JONE

Sample Detail

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BSK Log #: Sample Seal: Containers:

Sampled by: AR. HILLS NAVAKKO Reling	quished by: (Lithur Willard)	Date: 71/193 Time: 10:20 am pm
Received by: J. (4. Men.	Date: 21/1 93 Time: 10:25 am pm	
Laboratory: 35 M	Lab Reference: 1734	No of Samples:

Sample Detail

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Comments or Instructions:

BSK Log #: Sample Seals Containers:

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Laboratory: Lab Reference:	No of Samples:

Sample Detail

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Comments or Instructions:

7-129

TYPE: S. Due Date:

BSK Log #: Sample Seals: Containers:

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by: All Hurand Date:	: [02672 Time 16: 0 2 mm	erence:	
Hung NAVARRO Relinquished	Will Co Date:	Lab Reference	
Sampled by: ARt	Received by:	Laboratory:	

Sample Detail

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BSK Log #: Sample Sea... Containers: 101

Sample Activity Form

City of Fresno - Water Division, 1910 E. University Avc., Fresno, CA 93703

array Date: 6 125/ 93 Time: 3 . 47 .ms on		No of Samples:
- Relinquished by: Mith Murran Dale	Date: 102922 Time: 15:47 am pm	Lab Reference:
Sampled by: HRT NAVAKRO	Received by: () Mr. H, LLL()	Laboratory:

Sample Detail

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Comments or Instructions:

Sample Seals: P B Due Date: 7-77 Containers: B L

1222 Date: 6 1251 93 Time: 3 . 47 2011 pm		No of Samples:
Relinquished by: All Marine. Date:	Date: DM/PD Time: 15:出 am pm	Lab Reference:
Sampled by: ART GAYARRO	Received by: WHILLO	Laboratory:

Sample Detail

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Comments or Instructions:

TYPE: S Due Date:

Containers:

AVIARRO Relinquished by: Lett / Chroma Date: 6 1251 93 Time: 2 .47 mm and	10 Date: 125, 22 Time: 15: 47m pm	Lab Reference: No of Samples:
Sampled by: Arthya NAVIRERD	Received by: (1) ULO	Laboratory:

Sample Detail

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Comments or Instructions:

SSE Log #: Contrataure

City of Fresno - Water Division, 1910 E. University Ave., Fresno, CA 93703

Varia Dille: 6 124/93 Time: 3 :35 m our		No of Samples:
Sampled by: ART NAMEGO Relinquished by: Met Hyrang Date:	eived by: ALALICA COLLOCK Date: WH193 Time: 15:35 am pm	

Sample Detail

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Comments or Instructions:

Sample Stars (1)

City of Fresno - Water Division, 1910 E. University Ave., Fresno, CA 93703

400 mic 6 1241 93 Time: 1:00 my pm No of Samples: 104500 Date: 624193 Time: 13: 05 am pm Lab Reference: Relinquished by: Navarre Sampled by: A Received by: Laboratory:

Sample Detail

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City of Fresno - Water Division, 1910 E. University Ave., Fresno, CA 93703

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Sample Detail

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Comments or Instructions:

Sample Seals: A B L Date: 7/1998

City of Fresno - Water Division, 1910 E. University Ave., Fresno, CA 93703

Ture 13: 1 :0: 1 24/ 50 Time: 1 :0: 411 0111		No of Samples:
Sampled by: A MAN MAR Of Relinquished by: Ath Mutana Dane:	Received by Jacob Jack Dalla Date: 6124 93 Time: 13:05 am pm	Laboratory: / / / / Lab Reference:

Sample Detail

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Comments or Instructions:

Sample Seals: A P B L Date: THE

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: W-86	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	A.11.
Date =	$\frac{7/1}{}$ Time = $\frac{3.40}{}$
pH =	7.16
Temp. =	<u>22,7</u> deg C
Conductivity =	<u>0,55</u> mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	:
Temp. =	deg C
Conductivity =	mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: W-140	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	_a.M
Date =	<u>6/27</u> Time = <u>9:25</u> AM
pH =	7.12
Temp. =	23,4 deg C
Conductivity =	0,36 mS/cm
LOCATION: W-KV-DS	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	<u>Q.M.</u>
Date =	6/29 Time = 9:55 AM
pH =	7.46
Temp. =	23.3 deg C
Conductivity =	<u>0,28</u> mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-130 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Date = $\frac{6/28}{}$ Time = $\frac{2.145}{}$ P

pH = 7.28

Temp. = 25, 2 deg C

Conductivity = 0, 26 mS/cm

LOCATION: D-134 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = \mathcal{A} , \mathcal{M} .

Date = $\frac{6/28}{10}$ Time = $\frac{3:10}{10}$

pH = 7,50

Temp. = 26.7 deg C

Conductivity = 0.23 mS/cm

LOCATION: D-202 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 4.11.

Date = $\frac{6}{29}$ Time = $\frac{8:35}{35}$ Au

pH = 7.68

Temp. = 22.8 deg C

Conductivity = 0.23 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-8/ (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 1. M

Date = $\frac{\frac{1}{2}\sqrt{25/3}}{25}$ Time = $\frac{2.55}{2}$

DH = 7.24

Temp. = 29.7 deg C

Conductivity = 0.29 mS/cm

LOCATION: W-131 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _ A. 11.

Date = 428 Time = 1.55

pH= 7.42

Temp. = 24.4 deg C

Conductivity = 0.26 mS/cm

LOCATION: D-68 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = G. M.

Date = $\frac{6/28}{}$ Time = $\frac{2.15}{}$ P

pH= 7.13

Temp. = 24.9 deg C

Conductivity = 0.47 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-129 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _A. 11.

Date = $\frac{4/24/93}{2}$ Time = $\frac{2:30}{2}$

pH= 7.25

Temp. = 25.8 deg C

Conductivity = 0.28 mS/cm

LOCATION: D-3 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Date = $\frac{6}{2}\frac{4}{5}$ Time = 2:55

pH = 7.26

Temp. = 26. deg C

Conductivity = 0,35 mS/cm

LOCATION: W-91 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Date = 6/25/93 Time = 10:20 AM

pH = 7.63

Temp. = 22.3 deg C

Conductivity = 0.27 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-158 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = $\mathcal{A}\mathcal{M}$

Date = $\frac{10/25/93}{100}$ Time = $\frac{12:35}{9}$

 $pH = __{7.16}$

Temp. = 27.8 deg C

Conductivity = 0.35 ms/cm

LOCATION: D-117 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _ A . M

Date = $\frac{6/25/93}{105}$ Time = $\frac{1505}{105}$ P

pH = 7,32

Temp. = 28. / deg C

Conductivity = 0.31 mS/cm

LOCATION: D-155 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 9.4

Date = $\frac{6/25/93}{25}$ Time = $\frac{1.32}{25}$

pH = 7,23

Temp. = 29, 7 deg C

Conductivity = 0.75 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-156 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = <u>G.M.</u>

Date = $\frac{6/25/93}{1115}$ Time = $\frac{1115}{1115}$ AM

pH = 7.26

Temp. = 29.5 deg C

Conductivity = 0.28 mS/cm

LOCATION: D-128 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _\(\mathcal{U}\).

Date = $\frac{(1/25/9)^3}{(1/40)^2}$ Time = $\frac{11.40}{(1/40)^2}$

pH = 7. 21

Temp. = 26.2 deg C

Conductivity = 0.27 mS/cm

LOCATION: D-159 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = <u>A</u> M.

Date = 6/25/93 Time = 12 PM

DH = 7.22

Temp. = 25,7 deg C

Conductivity = 0.37 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-116 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = <u>A.M.</u>

Date = $\frac{6/25/93}{5}$ Time = $\frac{17:15}{5}$ PM

pH = 7.03

Temp. = 25.8 deg C

Conductivity = 0,36 mS/cm

LOCATION: D-69 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = \mathcal{A} . \mathcal{M} .

Date = $\frac{6/25/93}{25/93}$ Time = $\frac{ZPM}{2}$

pH = 7,03

Temp. = 26.7 deg C

Conductivity = 0:47 ms/cm

LOCATION: D-70 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Date = $\frac{6/25/9}{3}$ Time = $\frac{2.15}{2}$

pH= 7/12

Temp. = 29.7 deg C

Conductivity = 0.28 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = & M.

Date = $\frac{6/24}{}$ Time = $\frac{1:40}{}$

pH = 7.29

Temp. = 23.Z deg C

Conductivity = 0,33 ms/cm

LOCATION: D-18/ (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Date = $\frac{6/24}{1.55}$

 $pH = _{7.32}$

Temp. = 26.6 deg C

Conductivity = 0.31 mS/cm

LOCATION: D-79 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Date = $\frac{6/24/93}{1}$ Time = $\frac{2.15}{1}$

pH = 7.43

Temp. = 26.5 deg C

Conductivity = 0.27 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D- 66 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = <u> A. M</u>

Date = $\frac{16/23}{2}$ Time = $\frac{3.15}{2}$

DH = 7,47

Temp. = 25.4 deg C

Conductivity = 0.2/ mS/cm

LOCATION: D-76 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = \mathcal{G} . \mathcal{M} .

Date = $\frac{6/23}{1}$ Time = $\frac{3:30}{1}$

pH = 7,30

Temp. = $24.5 \deg C$

Conductivity = 0.23 mS/cm

LOCATION: D-126 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _ A. M.

Date = $\frac{6/24}{}$ Time = $\frac{10.00}{}$

DH = 7.74

Temp. = 23.9 deg C

Conductivity = 0.29 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: $D^{\#}3/$ (W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = A. 11.

Date = $\frac{6/23}{}$ Time = $\frac{2:25}{}$

 $pH = __{7.5/}$

Temp. = 25.8 deg C

Conductivity = 0.36 mS/cm

LOCATION: D-34 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = D=34 9. M.

Date = $\frac{C/23}{}$ Time = $\frac{2.45}{}$

pH = 7,68

Temp. = 28.6 deg C

Conductivity = 10.30 mS/cm

LOCATION: D-82 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _ A 1/1.

Date = $\frac{6/23}{}$ Time = $\frac{3.00}{}$

pH = 7.34

Temp. = 25.6 deg C

Conductivity = 0.28 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: 6-99 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = Omm

Date = 6/23 Time = 0830

pH = 7.53

Temp. = 22.1 deg C

Conductivity = 0.27 mS/cm

LOCATION: #133 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 9.11.

Date = $\frac{\ell/23}{}$ Time = $\frac{0845}{}$

DH = 7,45

Temp. = 25.4 deg C

Conductivity = 0.25 mS/cm

LOCATION: W- 97 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = \mathcal{U} . \mathcal{N} .

Date = $\frac{6}{23}$ Time = $\frac{9:55}{}$

pH = 7.36

Temp. = 22.9 deg C

Conductivity = 0.30 mS/cm

TABLE 5. Distribution System Sample Locations

Distribution System Sample Number	Distribution System Location Number	Address	Associated Pump Station Number
1	4/2479		91
2	6/24 129		79, 97
3	6/24126		KV-DS
4	L/28 68		86, 131
5	6/24 70		86
6	6/24 116		140
7	6/231336/23*		99
8	6/24 181 4/24		97
9	6/24 31 295		General* (91)
10	1/28/130		General* (91)
11	4/2366		General* (KV-DS)
12	.424159		General* (86, 140)
13	4/24158		General* (140)
14	6/28134		General* (99)
15	6/2334 5472		General* (99)
16	6/23 36 365		General* (99)
17	6/29202		General* (97)
18	6/24 81		General* (97)
19	6/24155		General* (131)
20	6/2469		General* (131)
21	6/23 82 ¥:		General**
22	424156		General**
23	b/24128		General**
- 24	6/24117		General**
25	6/2376		General**

^{*} General distribution system sample location, number in () represents remotely associated pump station.

^{**} General distribution system sample located in the same region as the tap sample locations.

TABLE 3. Source Water Sample Locations

	Pump Station Number	Address	Location	Associated Tap Sample Numbers
6/25	₩91	(b) (9)	(b) (9)	1,3
4/24	79			2
6/29	KV-DS			4,144,158
	86			10
0/29	* 140			Tract B*
6123	99			Tract C*
6 23	97			Tract D*
6/26	131			Tract E*

^{*} Tract B contains 63 sample sites, Tract C contains 46 sample sites, Tract D contains 29 sample sites, and Tract E contains 17 sample sites.

* Do not sample between room + 6 PM

TABLE 4. Source Sample Analytical Methods

Compound	EPA Method	Units	DLR*	Sample Container
Copper	200.7	mg/L	0.010	1 L polyethylene
Lead	239.2	mg/L	0.001	taken with copper
Calcium	200.7	mg/L	0.1	200 mL polyethylene
Alkalinity	310.1	mg/L as CaCO,	1.0	100 mL polyethylene

^{*} Detection limit for the purposes of reporting

	+		S	دو	1	#	Sample	aboratory:	sample ecciv
100			1/23/93	9/23/93	9/23/93 6:15AM AD	Date	Sample Detail	tory:	Sampled by: ARThur Received by: Cerit
			6:30 AM AD	6: Sam AD	6:15Am	Time		35/1	e Sie
				A D	AD	Туре			Varia
			i.			Code			MAMARRO
			# 173	エーント/ #	#/42-1	Location Description	90 =	Lab Reference:	MANARO Rollinguished by: Met Museum. Dale:
			X :	×	× #	2. 4 5 7 8 5 5 1 7 0 N F C M 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7		No of Samples:	52 m (m)

Comments or Instructions:

Sample Scale: P Containers: P

TYPE: S

Sampled by: Are to	MAYARRO MANAGON	1 12/1 1	Relinquished by: Date: 9/22/93 Time: 15:15 am pm No of Samples:
Sample Detail			No of San
# Date Time	Туре	Code	
1 9/22/93 8 AM	MAD		
2 9/22/93 6:15 AM AD	AM AD		# 148 841 #
	AM AD :	* 15	#
4 9/22/93 6 AM	N AD		
1 5.		_	
omments or Instructions:		_	

Sample Conta...

QUICK MEMO

DATE: August 5, 1993

TO: WTH, DLT, WEB, DSK, DEP, WJD *

MIZM

FROM: MARTIN R. MCINTYRE

SUBJECT: LEAD/COPPER SAMPLE RESULTS

As previously reported, our first round of sampling resulted in a 90th percentile lead value of 2.5 ppb- very good as the corrosion control optimization kicks in if the rise from source to tap is >than 5 ppb.

The Regulation required us to do a second round of sampling from the same taps six months later. Enter the accelerator of all chemical reactions- increased temperature. This second round collected in june/july resulted in a 90th percentile of 8 ppb, triggering a requirement to implement corrosion control optimization.

Given the 8 ppb results, the next step required by the regulation is a bench scale corrosion study. I have Discussed with HDR-they believe that there is a good chance that EPA will accept a "desk top" study which could demonstrate that the 8 ppb is equivalent to optimized corrosion control. The underlying assumption is that we can show that existing corrosion control programs in place elsewhere can't do any better than our existing condition. I'm reserving judgement on viability of this approach until after we meet with HDR- probably the week of Aug 15. Existing scope of work with HDR covers work through a bench scale corrosion study and includes the "desk top".

Bottom line- all is not well in the hinterland. May still dodge the corrosion control/lead bullet but its not assured.

Date: 8/2/93 To: Bell Dunn From: Art Museus Subject: Lead & Copper Sample Taken From Pump Station # 86 A. Status of Well trion tollection of Sample (Information From Don Burns) - Well was off for 3 weeks for repairs - Rump + Motor were pulled + repaired - Well was super Chlorinated - Acid was also added to Well consing - Well now for 5-10 - minutes prior to collection of sample B. Eurent Status of Well, Since Collections - Well is still not of line

- Has not passed Bae-T test

- Has Plate count of 3000

- Hero, been, super chlorinated several time

- Metal flake and seen during flushing Note: Decord sample may also not, be valid due to chamica I toutment

Date:		June 17, 1993		
Number	of Pages	(including cover page)	2	
Job No.	7200-011	fi.		



Message To

Name	Bill Dunn	Fax number called 209-488-1024
Firm	Water Division	
City	Fresno	State CA

If you did not receive all the pages, please call (916) 939-4100 as soon as possible. FAX No (916) 939-4143

From

Name	DAVE MURBACH		Department	141	
Autoatte	ndant Number	(916) 939-4100	Ext.	261	

Bill:

Attached is my schedule for activities next week. I plan to arrive at about 9:00 on Monday so that we can start delivering sample kits by 10:00. The schedule isn't all that rigid. I plan on getting back to your office between 3:30 and 4:00 each day so I can bring you up to speed.

Also I'd like to schedule a meeting for either 11:00 or 1:00 with you (and Martin, if he's not too busy) on Thursday so I can let you know what's left to be done of Friday.

Call me if you have any questions. Thanks.

Day

DMM schedule for week of June 21-24

			Time of D	ay	
Date	7:30 - 9:30	9:30 - 11:30	12:00 - 2:00	2:00 - 4:00	4:00 - 6:00
21-Jun	arrive In Fresno*, City pickup sample bottles	deliver sample kits	deliver sample kits	deliver sample kits	call @ sample bottles delivered
22-Jun	pickup samples	deliver samples to lab	deliver sample kits	deliver sample kits	call @ bottles delivered and no samples from 6/21
23-Jun	pickup samples	deliver samples to lab	distribution and source samples	distribution and source samples	call @ bottles delivered and no samples from 6/22
24-Jun	pickup samples & deliver to lab	call @ no bottles from 6/23	meet with Martin & Bill**	return to Sacramento	

^{*} Arrive in Fresno by 9:30

^{**} Meet at 11:00 if better, need to discuss work in remainder of week.

	E TAKE THIS W.		R CITY OF FRESNO'S WATER DIVISION TOMORROW
	HAVE ANY QUEST AT 498-4136.	TIONS OR CANNOT	TAKE THE SAMPLE FOR ANY REASON PLEASE CALL BILL
THANK	C YOU.		
	1		DIRECTIONS IPLE COLLECTION PROCEDURES
The co	llection procedure is	s described in detail b	pelow.
1.		ngs upon returning h	re sampling. The water department recommends that either early ome from work are the best sampling times to ensure that the
2.	the faucet and gen	tly open the cold wat	et is to be used for sampling. Place the open sample bottle below er tap. Slowly fill the sample bottle to the base of the neck and 45 seconds to fill the bottle.
3.	the form below, sig	n the form, and place	in the plastic bag provided. Fill in the information requested on in the plastic bag along with the sample bottle. Please review the re that all information contained on the label is correct.
4.	Place the sample of	outside your home for	pick-up by 9:00 in the morning.
5.	Results from this m for the State.	nonitoring effort will b	e provided to participating customers when reports are generated
Please	call Bill Dunn at 49	8-4136 if you have an	y questions regarding these instructions.
TO BE	COMPLETED BY	RESIDENT:	
PLEAS	E RETURN THIS	SHEET WITH THE	SAMPLE BOTTLE.
Water	was last used:	TIME:	DATE:
Sample	was collected:	TIME:	DATE:
I have i	read the above direc	tions and have taken	a tap sample in accordance with these directions.
Signatu	re		DATE:
Name:			
TO BE	COMPLETED BY	THE WATER DEPAR	RTMENT EMPLOYEE:
Sample 07200011.008	picked up by:		DATE:



Mr. Patrick Chan U.S. Environmental Protection Agency 75 Hawthorne Street San Francisco, California 94105-3901

Subject:

City of Fresno - Lead and Copper Rule, First Round Monitoring Results -

Preliminary Report.

Dear Patrick:

The intent of this letter is to transmit to you a summary of the results of the tap water sample analysis and the first set of source and distribution system water quality analysis. A final report detailing the results will follow after the results for the second round of water quality sampling are received. These results are expected by the middle of June.

The City of Fresno sampled tap water from 131 homes for lead and copper. All the homes were from a Tier 1-C sample pool. In addition to the above sampling, water quality samples were collected from 25 distribution system sites and eight pump stations identified as the source water for the residences sampled. The attached technical memorandum summarizes these results.

The City of Fresno will be conducting the second round of the initial tap water and water quality monitoring in late June 1993.

If you have any questions concerning these results please call me.

Very Truly Yours,

Perri Standish-Lee

Vice President

cc: Martin McIntyre - City of Fresno

Doug Kirk - City of Fresno Bill Dunn - City of Fresno Matt Bullis - City of Fresno

Cindy Forbes - California Department of Health Services

Steve Reiber - HDR Engineering, Inc. Dave Murbach - HDR Engineering, Inc.

TECHNICAL MEMORANDUM CITY OF FRESNO LEAD AND COPPER RULE COMPLIANCE MONITORING PRELIMINARY RESULTS OF INITIAL MONITORING - FIRST ROUND

The City of Fresno collected their first round of initial monitoring samples in compliance with the Lead and Copper Rule during April and May, 1993.

TAP WATER SAMPLE RESULTS

Table 1 presents the results of the tap water analysis for lead and copper. The City of Fresno sampled 131 single family residences that comprised a Tier 1-C sample pool. The table lists the lead and copper concentrations in descending order. This was done in order to determine the 90th percentile levels as required by the Lead and Copper Rule.

Lead Results

The 90th percentile lead level was determined by multiplying the number of samples taken by 0.9 (131 x 0.9 = 117.9). The 90th percentile lead level for the City of Fresno samples is 0.0025 mg/L which is below the EPA action level of 0.015 mg/L. It should be noted that the laboratory reports all analyzed lead values between 0.0010 and 0.0049 mg/L as 0.0025 mg/L.

Copper Results

The 90th percentile copper level was determined in the same way as for lead. The 90th percentile copper level for the City of Fresno is 0.43 mg/L which is below the EPA action level of 1.3 mg/L.

DISTRIBUTION SYSTEM AND POINT OF ENTRY SAMPLE RESULTS

Water quality analysis was performed on 25 distribution system locations and 8 points of entry to the distribution system. The results of this analysis are summarized in Tables 2 and 3. These results are the first set of two sets of sampling that will be performed in conjunction with the first round of initial monitoring.

The lead concentration of the source water was less than 0.001 mg/L at all sample locations.

FOLLOW-UP MONITORING

To comply with the requirements of the Lead and Copper Rule a second round must be conducted to complete the initial monitoring period. This second round of sampling will begin during the last week of June 1993.

07200011.010

TABLE 1. TAP WATER SAMPLE ANALYSIS

		Sample	Lead		Sample	Copper
Ranking		No.	mg/L		No.	mg/L
131		169	0.037		153	0.74
130		154	0.011		159	0.60
129		157	0.008		18	0.56
128		10	0.007		54	0.53
127		18	0.006		174	0.52
126		151	0.006		137	0.51
125		90	0.005		156	0.51
124		1	0.0025*		175	0.51
123		2	0.0025*		21	0.50
122		3	0.0025*		128	0.48
121		11	0.0025*		1	0.47
120		41	0.0025*		166	0.47
119		48	0.0025*		171	0.45
118	90th %	53	0.0025*	90th %	80	0.43
117		54	0.0025*		151	0.43
116		60	0.0025*		146	0.42
115		64	0.0025*		152	0.42
114		65	0.0025*		53	0.39
113		72	0.0025*		72	0.39
112		81	0.0025*		20	0.38
111		93	0.0025*		83	0.38
110		95	0.0025*		176	0.38
109		102	0.0025*		133	0.37
108		109	0.0025*		67	0.36
107		110	0.0025*		132	0.36
106		113	0.0025*		150	0.36
105		114	0.0025*		169	0.36
104		117	0.0025*		27	0.35
103		118	0.0025*		81	0.35
102		119	0.0025*		90	0.35
101		129	0.0025*		141	0.35
100		130	0.0025*		145	0.35
99		132	0.0025*		161	0.34
98		134	0.0025*		41	0.33
97		135	0.0025*		46	0.33
96		136	0.0025*		142	0.33
95		137	0.0025*		10	0.32
94		140	0.0025*		95	0.32
93		141	0.0025*		113	0.32
92		152	0.0025*	+	130	0.32
91		155	0.0025*		168	0.32
90		163	0.0025*	-	26	
89		166	0.0025*	+	148	0.31
88		168	0.0025*			0.31
87		170			88	0.30
86			0.0025*		93	0.30
		171	0.0025*	-	42	0.28
85		172	0.0025*		48	0.28

TABLE 1. TAP WATER SAMPLE ANALYSIS

Ranking	Sample No.	Lead mg/L	Sample	Copper
84	173	0.0025*	No.	mg/L
83	174	0.0025*	65	0.27
82	175	0.0025*	107	0.26
81	176	0.0025*		0.26
80	4		120	0.26
79	20	< 0.001	2	0.25
78	21	< 0.001	30	0.25
77	25	< 0.001	64	0.25
76	26	< 0.001	115	0.25
75	27	< 0.001	149	0.25
74	30	< 0.001 < 0.001	163	0.25
73	36		165	0.25
72		< 0.001	79	0.24
	38	< 0.001	75	0.23
71	39	< 0.001	102	0.23
70	40	< 0.001	114	0.23
69	42	< 0.001	51	0.22
68	43	< 0.001	62	0.22
67	46	< 0.001	39	0.21
66	50	< 0.001	63	0.21
65	51	< 0.001	124	0.21
64	52	< 0.001	109	0.20
63	55	< 0.001	55	0.19
62	56	< 0.001	76	0.19
61	57	< 0.001	100	0.19
60	58	< 0.001	119	0.19
59	61	< 0.001	131	0.19
58	62	< 0.001	136	0.19
57	63	< 0.001	170	0.19
56	66	< 0.001	71	0.18
55	67	< 0.001	78	0.18
54	69	< 0.001	112	0.18
53	71	< 0.001	4	0.17
52	74	< 0.001	40	0.17
51	75	< 0.001	50	0.17
50	76	< 0.001	89	0.17
49	78	< 0.001	110	0.17
48	79	< 0.001	143	0.17
47	80	< 0.001	164	0.17
46	83	< 0.001	25	0.16
45	84	< 0.001	125	0.16
44	86	< 0.001	38	0.15
43	87	< 0.001	157	0.15
42	88	< 0.001	173	0.15
41	89	< 0.001	56	0.14
40	91	< 0.001	58	0.14
39	92	< 0.001	66	0.14
38	98	< 0.001	87	0.14

TABLE 1. TAP WATER SAMPLE ANALYSIS

Ranking	Sample No.	Lead mg/L	Sample No.	Copper mg/L
37	100	< 0.001	118	0.14
36	103	< 0.001	167	0.14
35	105	< 0.001	91	0.13
34	107	< 0.001	105	0.13
33	108	< 0.001	108	0.13
32	111	< 0.001	122	0.13
31	112	< 0.001	3	0.12
30	115	< 0.001	98	0.12
29	116	< 0.001	129	0.12
28	120	< 0.001	139	0.12
27	122	< 0.001	172	0.12
26	123	< 0.001	43	0.11
25	124	< 0.001	135	0.11
24	125	< 0.001	57	0.10
23	126	< 0.001	60	0.10
22	127	< 0.001	92	0.10
21	128	< 0.001	116	0.10
20	131	< 0.001	144	0.10
19	133	< 0.001	36	0.09
18	139	< 0.001	61	0.09
17	142	< 0.001	111	0.09
16	143	< 0.001	127	0.09
15	144	< 0.001	86	0.08
14	145	< 0.001	155	0.08
13	146	< 0.001	126	0.07
12	147	< 0.001	134	0.07
11	148	< 0.001	147	0.07
10	149	< 0.001	103	0.06
9	150	< 0.001	117	0.06
8	153	< 0.001	123	0.06
7	156	< 0.001	52	0.025**
6	158	< 0.001	69	0.025**
5	159	< 0.001	74	0.025**
4	161	< 0.001	84	0.025**
3	164	< 0.001	140	0.025**
2	165	< 0.001	158	0.025**
1	167	< 0.001	154	< 0.01

^{*} The laboratory reports lead values between 0.0010 and 0.0049 mg/L as 0.0025 mg/L.

^{**} The laboratory reports copper values between 0.010 and 0.049 mg/L as 0.025 mg/L.

TABLE 2. DISTRIBUTION SYSTEM WATER QUALITY ANALYSIS

				Field Analys	is			
Sample No.	Alkalinity (mg/L CaCO3)	Calcium (mg/L)	pН	Temp. (deg C)	Cond. (mS/cm)	Date Sampled	Time Sampled	Sample
D-31	140	29	7.19	22.4	0.34	5/3/93	12:30	JAM
D-34	120	26	7.58	22.7	0.29	4/30/93	14:55	JAM
D-36	110	27	7.44	22.8	0.30	4/30/93	15:15	JAM
D-66	91	18	7.48	20.8	0.20	4/29/93	10:45	JAM
D-68	160	37	7.15	21.8	0.38	4/29/93	12:30	JAM
D-69	140	35	7.22	23.6	0.35	4/30/93	14:00	JAM
D-70	93	20	7.56	21.5	0.23	4/29/93	15:10	JAM
D-76	110	22	7.56	20.6	0.25	5/4/93	9:25	JAM
D-79	97	23	7.54	22.1	0.26	5/3/93	10:55	JAM
D-81	110	28	7.38	23.5	0.31	5/3/93	10:05	JAM
D-82	110	26	7.63	22.8	0.28	5/3/93	15:45	JAM
D-116	140	34	7.22	24.3	0.33	5/3/93	14:20	JAM
D-117	130	33	7.39	23.6	0.32	5/3/93	15:00	JAM
D-126	79	17	7.55	22.5	0.19	4/29/93	9:55	JAM
D-128	170	35	7.10	22.5	0.36	5/3/93	13:00	JAM
D-129	110	27	7.33	25.4	0.31	5/3/93	9:17	JAM
D-130	110	23	7.41	22.4	0.27	5/3/93	11:15	JAM
D-133	120	25	7.47	21.0	0.28	4/22/93	11:10	DMM
D-134	110	26	7.46	23.8	0.28	4/30/93	15:40	JAM
D-155	81	17	7.44	23.4	0.21	4/29/93	13:15	JAM
D-156	130	25	7.27	23.2	0.29	5/3/93	14:45	JAM
D-158	120	25	7.37	24.0	0.29	5/3/93	14:00	JAM
D-159	130	26	7.17	23.5	0.31	4/29/93	15:25	JAM
D-181	120	27	7.44	22.9	0.32	5/3/93	9:45	JAM
D-202	120	28	7.34	23.0	0.31	5/3/93	9:30	JAM

Average	116	26	7.43	22.2	0.29
Minimum	91	18	7.15	20.6	0.20
Maximum	160	37	7.63	23.6	0.38

TABLE 3. SOURCE WATER QUALITY ANALYSIS

						Field Analysis	s	ALTERNATION OF THE PERSON OF T		
Sample	Lead	Copper	Alkalinity	Calcium	Hd	Temp.	Cond.	Date	Time	
No.	(mg/L)	(mg/L)	(mg/L CaCO3)	(mg/L)		(deg C)	(mS/cm)	Sampled	Sampled	Sampler
W-79	<0.001	<0.01	130	28	7.43	23.1	0.30	5/3/93	10:20	JAM
W-86	<0.001	<0.01	130	30	7.30	23.1	0.31	4/29/93	14:50	JAM
W-91	<0.001	<0.01	66	23	7.74	21.4	0.26	5/3/93	10:35	JAM
W-97	<0.001	<0.01	110	27	7.32	21.9	0.30	5/3/93	9:02	JAM
W-99	<0.001	0.025*	89	17	7.57	21.5	0.21	4/22/93	10:45	DMM
W-131	<0.001	<0.01	97	19	7.41	24.0	0.23	4/29/93	11:15	JAM
W-140	<0.001	<0.01	110	31	7.15	31.8	0.35	4/28/93	13:55	JAM
W-KVDS	<0.001	<0.01	79	16	7.62	22.5	0.20	4/29/93	10:17	JAM

Average	<0.001	0.003	106	24	7.44	23.7	0.27
Minimum	<0.001	<0.01	79	16	7.15	21.4	0.20
Maximum	<0.001	0.025*	130	31	7.74	31.8	0.35

* The laboratory reports copper values between 0.010 and 0.049 mg/L as 0.025 mg/L.

Date:	May 27, 1993		
Number	of Pages (including cover page)	2	
Job No.	7200-011		



Message To

Name Bill Dunn	Fax number called 209-488-1024
Firm Water Division	
City Fresno	_StateCA

If you did not receive all the pages, please call (916) 939-4100 as soon as possible. FAX No (916) 939-4143

From

Name Dave Munbaci	1	Department	141	
_Autoattondant Number	(916) 939-4100	Ext.	261	

Bill:

Attached is the letter that needs to go to the 130 residences sampled during the first round of sampling to inform them when the second round of sampling will occur.

In the past we've followed the steps below to get the letter sent out to the customers:

- Jeff and Martin review the letter and make any minor changes as needed
- The letter is retyped on the City's letterhead and signed by Martin
- The letter is sent back to HDR along with the City's envelopes
- HDR copies the original letter, creates the address labels, and mails out all the letters

In this case I will need at least 130 envelopes to mail these letters and if I get the original letter back by next Tuesday, they will all be mailed by June 2.

Call me if you have any questions. Thanks.



Dear Resident:

Thank you for participating in our lead and copper tap water monitoring program. The success of this program has been dependent upon people like you who volunteered to collect the residential tap water samples according to the requirements of the U.S. Environmental Protection Agency (EPA) for the Lead and Copper Rule.

A total of 130 tap water samples were collected from different residences in the City of Fresno's service area between April 21 and April 30, 1993. A preliminary review of the analytical results from these samples shows that the results for both lead and copper are reassuringly very low. You will be receiving your individual results after we have completed the second round of lead and copper tap water monitoring.

The EPA Lead and Copper Rule requires a second round of monitoring to verify the initial results, and with your permission, we would like to receive another sample from your home during the week of June 21-25. We will be delivering sample bottles and instructions on Monday, Tuesday, and Wednesday of that week. Please follow the instructions again and collect the sample by the morning after you receive your sample bottle, if possible. If you will not be available during this week, please call Bill Dunn at 498-4136 so we can reschedule your sample collection for either the previous or following week for your convenience.

Thank you again for participating in our lead and copper monitoring program. If you have any questions or comments please call Bill Dunn at 498-4136.

Sincerely,

Martin R. McIntyre Water Systems Supervisor II



Mr. Patrick Chan U.S. Environmental Protection Agency 75 Hawthorne Street San Francisco, CA 94105-3901

Subject:

City of Fresno - Lead and Copper Rule Sampling Plan

Dear Patrick:

The City of Fresno initiated their first round of lead and copper monitoring the week of April 19, 1992.

Our material survey concluded that no Tier I-A, or Tier I-B sites could be located in the City of Fresno Service Area. We therefore included in the sampling pool Tier I-C sites only. The first round of samples will include 130 homes. Table 1 is a list of the homes sampled. The City identified eight wells, Table 2, which serve the areas sampled. These wells were sampled and will be analyzed for: lead, copper, pH, conductivity, calcium, alkalinity and temperature.

In addition, twenty five distribution system sampling sites, Table 3, were identified in and around the homes participating in the study for water quality monitoring. These stations were sampled and will be analyzed for: pH, conductivity, calcium, alkalinity and temperature.

Source water quality and distribution system quality will be sampled again in mid-May to comply with the Lead and Copper Rule monitoring requirements.

The City will initiate the second round of sampling the last week of June. We should have all the samples to the laboratory by July 11, 1993. A copy of our sample plan is included for your review.

As soon as we receive the results from the laboratory, we will prepare a report which summarizes the results. A copy of the report will be sent to you and to the California Department of Health Services.

Mr. Patrick Chan April 30, 1993 Page 2

If for some reason the sampling plan does not meet with your satisfaction or you have any coments about our sampling plan, please contact me as soon as possible, so we can make the necessary changes.

Very truly yours,

Perri Standish-Lee

Peuri Standish-Lee

Vice President

PSL/af

cc:

Martin McIntyre

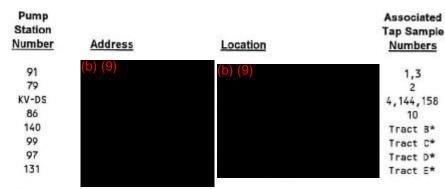
Bill Dunn

Doug Kirk

Matt Bullis

Cindy Forbes

TABLE 2. Fresno Lead and Copper Rule **Source Sample Locations**



Source Sample : Lead, Copper, pH, Temp., Cond., Calcium, Alkalinity Analysis

* The

	ts represent the sample	numbers taken from eac	h tract.
Tract B	Tract C	Tract D	Tract E
11	25	20	21
18	30	26	27
127	36	39	41
128	38	43	42
129	40	46	48
130	52	50	51
131	53	55	54
132	56	65	64
133	57	72	67
134	58	75	80
135	60	76	81
136	61	84	82
137	62	88	83
138	63	90	95
139	66	92	107
140	69	93	113
141	71	96	164
142	74	114	
143	78	115	
145	79	116	
146	86	117	
147	87	118	
148	89	119	
149	91	120	
150	98	163	
151	99		
152	100		
153	102		
154	103		
155	105		
156	108		
157	109		
159	110		
160	111		
161	112		
166	122		
167	123		
168	124		
169	125		
170	126		
171	165		
172			
173			
174			
175			

176

TABLE 3. Fresno Lead and Copper Rule Distribution System Sample Locations

System Sample	System Location		Associated Pump Station
Number	Number	Address	Number
1	79 (b	(6)	91
2	129		79, 97
3	126		KV-DS
4	68		86, 131
1 2 3 4 5	70		86
6	116		140
7	133		99
8	181		97
9	31		General* (91)
10	130		General* (91)
11	66		General* (KV-DS)
12	159		General* (86, 140)
13	158		General* (140)
14	134		General* (99)
15	34		General* (99)
16	36		General* (99)
17	202		General* (97)
18	81		General* (97)
19	155		General* (131)
20	69		General* (131)
21	82		General**
22	156		General**
23	128		General**
24	117		General**
25	76		General**

^{*} General distribution system sample location , number in () represents remotely associated pump station.

Distribution Sample : pH, Temperature, Conductivity, Calcium, Alkalinity Analyses

^{**} General distribution system sample located in the same region as the tap sample locations.

SAMPLE PLAN INITIAL MONITORING

FIRST DRAW TAP SAMPLES

The initial monitoring period consists of two consecutive 6-month sample periods during which the water is sampled from the homeowner's tap and analyzed for total lead and copper. The first set of these samples is to be collected between April 20 and 30, 1993. The second set of samples is expected to be collected in July 1993.

A minimum of 100 samples must be taken for each set. Table 1 presents a list of 136 homeowners that volunteered to collect samples at their tap. The instructions the homeowner will use to collect the samples are presented in Figure 1. An attempt should be made to hand deliver the sample bottles and demonstrate the sampling technique to as many homes as possible. Also, the presence of lead solder and copper pipe must be verified in 10-15% of the homes before the first set of samples can be taken.

The sample collection procedures will follow all EPA Guidelines in the Lead and Copper Rule. Some of the important sampling criteria include:

- The sample must be collected from a kitchen or bathroom tap.
- The water must stand in the pipes for at least 6 hours before the sample is collected.
- The sample must be a first-draw collection.
- The sample must be 1 liter in volume.

Sample kits should be prepared which include a 1 liter polyethylene sample bottle and written instruction (which includes a chain of custody form). The sample kits should be assembled in a 1 gallon plastic bag. Prior to delivery of the sample kits a label should be placed on each set of written instructions that includes the sample number and the

address. This sample number must also be entered on the sample bottle label.

The sample kits will be delivered to homeowners in person, if available, after again ascertaining that the home has copper plumbing. The homeowner will be instructed in taking the sample and how to fill out the chain of custody. Sample kits can be left at a homeowners doorstep if they cannot be reached in person. A follow up call to these homeowners should be made to make sure they understand the instructions.

The residents will then collect the sample and leave the sample bottle outside their home for pick-up the next morning. HDR and City Staff will pick-up the samples, verify that the sample number matches the address, and deliver the samples to BSK Laboratories, Inc. for analysis (see table below). A chain of custody must be filled out for all samples before delivery to the laboratory.

Lead and Copper Analytical Methods

Compound	EPA Method	Units	DLR*
Copper	200.7	mg/L	0.010
Lead	239.2	mg/L	0.001

^{*} Detection limit for the purposes of reporting

QA/QC samples will be given to the laboratory along with the homeowner's samples. These samples will include spiked samples with a known concentration of lead, blank samples of distilled water, and duplicate samples split from spiked samples. These samples will represent approximately 10% of the total samples collected.

SOURCE WATER SAMPLES

The initial monitoring period consists of two consecutive 6-month sample periods during which the water is sampled from the well system. These samples are analyzed for lead, copper, pH, conductivity, temperature, alkalinity, and calcium. The first set of these samples is to be collected between April 20 and 30, 1993 along with the first draw tap samples. The second half of this set of samples is expected to be collected the week of May 17, 1993. The second set of samples will be collected the weeks of June 20 and July 12, 1993.

The source water for each first draw sample site must be sampled. Table 2 presents a list of eight well sites which are in the same general area as the first draw tap sample sites.

To collect the well samples, first flush the sample tap. Then collect a sample for field analysis of pH, temperature, and conductivity. Record the field measurements on a log sheet including the date, time, and location of sample. After performing the field analysis, collect the samples for lead, copper, alkalinity, and calcium in sample bottles provided by BSK Laboratories. Cool the samples for calcium and alkalinity to 4° C, label the bottles with the sample location, complete a chain of custody form for these samples, and deliver them to the laboratory within 24 hours for analysis (see table below).

Source Sample Analytical Methods

Compound	EPA Method	Units	DLR*
Copper	200.7	mg/L	0.010
Lead	239.2	mg/L	0.001
Calcium	200.7	mg/L	0.1
Alkalinity	310.1	mg/L as CaCO ₃	1.0

^{*} Detection limit for the purposes of reporting

DISTRIBUTION SYSTEM SAMPLES

The initial monitoring period consists of two consecutive 6-month sample periods during which the water is sampled twice from the distribution system. These samples are analyzed for pH, conductivity, temperature, alkalinity, and calcium. The first set of these samples is to be collected between April 20 and 30, 1993 along with the first draw tap samples. The second half of this set of samples is expected to be collected the week of May 17, 1993. The second set of samples will be collected the weeks of June 20 and July 12, 1993.

A minimum of 25 samples must be taken for each set. Table 3 presents a list of 25 coliform sample sites which are in the same general area as the first draw tap sample sites.

To collect the distribution system samples, first flush the sample tap as you would to collect a coliform sample. Then collect a sample for field analysis of pH, temperature, and conductivity. Record the field measurements on a log sheet including the date, time, and location of sample. After performing the field analysis, collect the samples for alkalinity and calcium in sample bottles provided by BSK Laboratories. Cool the samples to 4 C, label the bottles with the sample location, complete a chain of custody form for these samples, and deliver them to the laboratory within 24 hours for analysis (see table below).

Distribution System Sample Analytical Methods

Analysis	EPA Method	Units	DLR
Calcium	200.7	mg/L	0.1
Alkalinity	310.1	mg/L as CaCO ₃	1.0

Detection limit for the purposes of reporting

- Do not use any water for 6-8 hours before sampling. The water department recommends that either
 early mornings or evenings upon returning home from work are the best sampling times to ensure that
 the necessary water conditions exist.
- A kitchen or bathroom cold-water faucet is to be used for sampling. Place the open sample bottle
 below the faucet and gently open the cold water tap. Slowly fill the sample bottle to the base of the
 neck and turn off the water. It should take about 45 seconds to fill the bottle.
- 3. Tightly cap the sample bottle and place in the plastic bag provided. Fill in the information requested on the form below, sign the form, and place in the plastic bag along with the sample bottle. Please review the address label below at this time to ensure that all information contained on the label is correct.
- Place the sample outside your home for pick-up by 9:00 in the morning.
- Results from this monitoring effort will be provided to participating customers when reports are generated for the State.

MEETING AGENDA

May 25, 1993

City of Fresno, Water Division

First Round Initial Monitoring Status

- 130 samples taken
- 126 analyzed to date
- 90th percentile levels for lead (0.0025mg/L) and copper (0.42 mg/L)
- last for samples taken on 5/12-13 and results should be available soon
- only one sample above 11ppb lead at 37ppb

Results to Consumers

- example letter prepared
- letter is individual to each consumer, need to get Fresno letterhead to print the letters, signatures?

Results to EPA

- preliminary summary report drafted for Patrick Chan
- final detailed report after water quality analysis finished for first round of monitoring program (mid-June)

Future Monitoring

- Second water quality samples for first round of initial monitoring during this week (distribution and pumping station sites only)
- Second round of initial monitoring during week of June 21
- Second water quality samples for second round of initial monitoring during week of July 12 (distribution and pumping station sites only)

Materials Survey

- Draft Technical Memo No. 1
- Summarizes and documents the activities of the material survey to the point where the addresses 538 potential Tier 1 sample sites are presented and 162 volunteers from these sites were solicited
- Missing letter from City's distribution department stating that lead lines are replaced as found and the neighbors homes are checked, no known lead lines exist, and when was the last time lead lines were found
- Missing letter from the City to homes initially accepted into the monitoring program but were found to have no lead solder upon field testing

Sampling Plan

- Draft Technical Memo No. 2
- Summarizes the procedures used to select the tap water sample sites as
 well as the water quality sample sites; procedures for collecting each type
 of sample and the analysis that must be performed is also included

Report Presentation Format

- In a single binder with the following potential divisions:

Material Survey
Sampling Plan
First Round Initial Monitoring Results
Second Round Initial Monitoring Results
Followup Monitoring Results
EPA / DHS Correspondence

819 water systems found high in lead

Unacceptable amounts reported in supplies that serve 30 million people.

By Rudy Abramson Los Angeles Times

WASHINGTON — Unacceptable amounts of lead have been detected in the public water supplies of 819 medium-sized and large American communities, which together serve 30 million people, the Environmental Protection Agency reported Tuesday.

The levels were monitored in residences in those areas deemed

to be at high risk for lead contamination and do not represent average amounts for those water systems, the EPA cautioned. Nevertheless, agency officials said Carol Browner



they were "very concerned" over the continuing threat of lead in water supplies, as well as in the air, soil and paint.

Under federal regulations, when lead in public water exceeds 15 parts per billion, officials must take steps to control water pipe corrosion, teach the public how to minimize the hazard and continue monitoring the supply.

"The reduction of children's exposure to lead is one of EPA's top priorities," EPA Administrator Carol Browner said. "While systems with elevated levels are required to reduce their lead levels through corrosion control measures, there are also important steps that consumers can take to help prevent exposure and increase safety."

Exposure to lead contamination is particularly damaging to infants and children - impairing hearing and eyesight and causing brain damage. It is estimated that one child in every six has elevated levels of lead in the blood. While officials say the source of greatest concern is house paint, about 20 percent of

The local story

Fresno is about halfway through a test sampling for lead content in city water, according to Martin McIntyre, city water quality chief.

He said the results so far have been well within the limits set by the Environmental Protection Agency. Fres-no has not had problems with lead in city water the said.

Tests are being conducted in accordance with EPA requirements, which include sampling water in the homes of customers. The corrosive action of water in delivery pipes is a common cause of lead in the water, McIntyre

Contamination has been a problem with city walls, however. More than 40 wells have been shut down since 1989 because agricultural contaminants were discovered.

cases are attributed to water supplies.

According to the study made public Tuesday, lead levels as high as 484 parts per billion were recorded at the Marine Corps' Camp Lejeune at Hadnot Point, N.C.

For the most part, the excessive levels were found in systems in the East, which tend to be older and in which lead was used to seal pipe joints.

Altogether, samples from 7,500 water systems have been analyzed. But James Elder, director of the EPA's office of ground water and drinking water, said 1,100 systems failed to conduct required monitoring, leaving them liable for \$5,000 federal fines.

Results of the initial survey of big-city systems — those serving more than 50,000 customers were made available in October. Since then, Elder said, many have begun to reduce levels.

TABLES FRESNO LEAD & COPPER RULE DISTRIBUTION SYSTEM SAMPLE LOCATIONS

Distribution System Sample <u>Number</u>	Distribution System Location Number	Date Sample <u>Taken</u>	Address	Acres	Associated Pump Station Number
1	79 3/20	1 5/3 (b) (6)			000
2	129 5/2	- July 3			91
3	126 5/2	4 4/29			79, 97
4	68 5/2	A procedure			KV-DS
5	70 5/24	The state of the s			86 131
2 3 4 5 6 7	1165/24	5/2			86
7	122 / /	4/22			140
8 9	181 5/26	Contraction of the Contraction o			99
	31 5/24	2 manual and a second			97
10	130 5/24	A Service and the service and			General* (91)
11	66 5/24	139			General* (91)
12	159 5/24	1/20			General* (KV-DS)
13	158 5/24	7/27			General* (85, 140)
- 14	134 5/26				General* (140)
15	34 5/2	7/30			General* (99)
16	36	6 7/30			General* (99)
17	200 3/26				General* (99)
18	84	5/3			General* (97)
19	155 5/26	3/3			General* (97)
20		4127			General* (131)
21	69 5/24	9/30			General* (131)
22	82 5/24	1 5/3			General**
23	156 5/24				General**
24	128 5/24	-			Generai**
25	117 5/24	5/3			General**
23	76 5/24	5/4			General**

^{*} General distribution system sample location , number in () represents remotely associated pump station.

Distribution Sample = pH, Temperature, Conductivity, Calcium, Alkalinity

Fax	1501-180-105+
Phone 4	P. P. celation
co. HO.2	Signery O sapace
From Dave Murball	17701219
Renne 7671 # of pages # 2	

^{**} General distribution system sample located in the same region as the tap sample locations.

TABLE 2. FRESNO LEAD & COPPER RULE SOURCE SAMPLE LIST

	Date Sample <u>Taken</u>	Pump Station Number	Address	Location	Associated Tap Sample <u>Numbers</u>
5/24 5/24 5/24 5/24 5/26 5/26 5/24	5/3 5/3 4/29 4/29 4/22 5/3 4/29	91 79 KV-DS 86 140 99 97	(b) (9)	(b) (9)	1,3 2 4,144,158 10 Tract B* Tract C* Tract D*

Source Sample = Lead, Copper, pH, Temp., Conductivity, Calcium, Alkalinity

^{*} The following lists represent the sample numbers taken from each tract.

Tract B	Tract C	Tract D	Tract
18	126	119	42
129	52	93	54
146	71	50	48
130	58	90	53
152	30	117	21
143	100	116	41
153	38	20	80
134	61	118	95
155	87	162	107
137	125	46	64
156	53	75	51
154	74	55	164
150	62	59	81
11	60	92	82
159	78	43	113
160	98	96	67
147	86	88	27
127	110	26	61
157	122	39	
148	69	72	
139	57	84	
145	91	120	
142	123	76	
128	111	163	
161	105	28	
136	103	70	
131	66	121	
140	112	35	
149	89	115	
141	63	114	
133	102		
135	99		
151	40		
138	56		
132	109		
	108		
	79		
	36		
	124		
	105		

165 25

- 1				_
	LOCATION:	(W-## =	WELL SITE)	
		(D-## =	DISTRIBUTION SYSTEM SITE)	
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	Temp. =	' <u></u>	_ deg C	
	Conductivity =		_ mS/cm	
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	Samplers Initials =			
	Date =		Time =	
	pH =			
	Temp. =		deg C	
	Conductivity =		mS/cm	
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	Temp. =		deg C	
	Conductivity =		mS/cm	

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Temp. =	deg C	
Conductivity =	mS/cm	
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Samplers Initials =		
Date =	Time =	
pH =	*	
Temp. =	deg C	
Conductivity =	mS/cm	

LOCATION:	(W-## = WELL SITE)
	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

LOCATION:	(W-## = WELL SITE)
	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

LOCATION:	(W-## = WELL SITE)	
	(D-## = DISTRIBUTION SYSTEM SITE)	
Samplers Initials =		
Date =	Time =	
pH =		
Temp. =	deg C	
Conductivity =	mS/cm	
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)	
Samplers Initials =		
Date =	Time =	
pH =		
Temp. =	deg C	
Conductivity =	mS/cm	
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)	
Samplers Initials =		
Date =	Time =	
pH =	<u> </u>	
Temp. =	deg C	
Conductivity =	mS/cm	

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The state of the s	pH =	
Conductivity = ms/cm	Temp. =	deg C
2010A3W3W4	Conductivity =	mS/cm

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	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
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Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
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Temp. =	deg C
Conductivity =	mS/cm
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Conductivity =	mS/cm
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Temp. =	deg C
Conductivity =	mS/cm

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Samplers Initials =	·
Date =	Time =
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	deg C
Conductivity =	mS/cm
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Samplers Initials =	
Date =	Time =
pH =	<u>*</u>
Temp. =	deg C
Conductivity =	mS/cm

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	Temp. =	deg C
	Conductivity =	mS/cm
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L	Conductivity =	mS/cm
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	Temp. =	deg C
	Conductivity =	mS/cm

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Temp. =	deg C
Conductivity =	mS/cm
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Date =	Time =
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Temp. =	deg C
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Temp. =	deg C
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	Temp. =	deg C
	Conductivity =	mS/cm
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	Temp. =	deg C
	Conductivity = _	mS/cm
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	LOCATION:	(W-## = WELL SITE)
		(D-## = DISTRIBUTION SYSTEM SITE)
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	Conductivity =	mS/cm
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	pH =	
	Temp. =	deg C
	Conductivity =	mS/cm
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	Temp. =	deg C
	Conductivity =	mS/cm
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	Date =	Time =
	pH =	
	Temp. =	deg C
	Conductivity =	mS/cm
	LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
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	Date =	Time =
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	Temp. =	deg C
	Conductivity =	mS/cm

	97.
LOCATION:	(W-## = WELL SITE)
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Samplers Initials =	
Date =	Time =
Temp. =	deg C
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LOCATION:	(W-## = WELL SITE)
Samplers Initials =	(D-## = DISTRIBUTION SYSTEM SITE)
	Time =
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	-
Temp. =	deg C

LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
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pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
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Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	*
Temp. =	deg C
Conductivity =	mS/cm

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Second to the factor of the fa	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	·
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
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Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	*
Temp. =	deg C
Conductivity =	mS/cm

	LOCATION:	_ (W-## = WELL SITE)
		(D-## = DISTRIBUTION SYSTEM SITE)
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	Date =	Time =
	pĤ =	·
	Temp. =	deg C
	Conductivity =	mS/cm
	LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
	Samplers Initials =	
	Date =	Time =
	pH =	
	Temp. =	deg C
	Conductivity =	mS/cm
	LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
	Samplers Initials =	
	Date =	Time =
	pH =	
	Temp. =	deg C
handari a	Conductivity =	mS/cm

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LOCATION:	(W-## = WELL SITE)
	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers In	itials =
	Date = Time =
	pH =
Te	emp. = deg C
Conduct	ivity = mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Init	tials =
	Date = Time =
	pH =
Te	mp. = deg C
Conducti	vity = mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
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Ten	np. = deg C
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Temp. =	deg C
Conductivity =	mS/cm
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Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	36
Date =	Time =
pH =	<u>*</u>
Temp. =	deg C
Conductivity =	mS/cm

LOCATION:	(W-## = WELL SITE)
	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	·
Date =	Time =
pĤ =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	-
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

LOCATION:	(W-## = WELL SITE)
	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
рН =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	-
Date =	Time =
	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH = _	E
	deg C
Conductivity =	mS/cm

LOCATION:	(W-## = WELL SITE)
	(D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	(49)
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	*
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	9
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

Complet be				
Sampled by:		-		
Received by:				/ Time:
Laboratory:_				an pm
Sample Detail				The Committee.
# Date	Time	Туре	Code	5 5 5 5 5 5 5 5 9 N F C 0 0 0 0 1 2 3 4 0 T Location Description 2 4 5 7 8 5 5 1 7 0 Y
-				
	+			
numents or Instructions:	tructions:			

Sampled by:			Relinquished by:	Date:	-/_/_ Ti	Tanc: :
Laboratory:			Date : / Time: _ Lab Reference:	_: am pn	No of Camples	1
Sample Detail					TWO OF Samples.	
# Date Time	Туре	Code	Location Description	5 5 5 5 5 0 0 0 0 0 2 4 5 7 8	5 5 5 5 9 1 2 3 4 0 5 5 1 7 0	H - K - H O
			249			
omments or Instructions:						

Sampled by:			T	Relinquished by:		
Received by:				Date :// Time:	1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1	am pm
aboratory:				1	No of Samples:	
Sample Detail						
# Date	Time	Туре	Code	Location Description	5 5 5 5 5 5 5 5 5 9 N F C 0 0 0 0 0 1 2 3 4 0 T 2. 4 5 7 8 5 5 1 7 0 Y	
numents or Instructions:	ione:		-			

Date:

May 25, 1993

Attendees:

Martin McIntyre (City of Fresno) Bill Burmeister (City of Fresno)

Doug Kirk (City of Fresno)
Jeff Murray (City of Fresno)
Bill Dunn (City of Fresno)
Matt Bullis (City of Fresno)

Perri Standish-Lee (HDR Engineering) Dave Murbach (HDR Engineering)

Location:

City of Fresno Water Division Offices

Subject:

Lead & Copper Rule Compliance Monitoring Project Status

NOTE: BOLD TEXT INDICATES ACTION ITEMS

I. STATUS OF FIRST ROUND OF INITIAL MONITORING PROGRAM

a. Tap Samples

- 130 samples were collected and 126 have been analyzed to date (the last four samples were collected on 5/12 and 5/13 and results are not yet available).
- The 90th percentile level for lead is 0.0025 mg/L.
- The 90th percentile level for copper is 0.42 mg/L.
- These results are well below the EPA action levels of 0.015 mg/L for lead and 1.30 mg/L for copper.
- Only one sample had a lead concentration greater than the action level (0.037 mg/L). Jeff M. collected a duplicate sample this morning to verify this high level. The resident also told Jeff that they had no new faucets.
- An example letter that will transmit the individual sample results to each resident in the monitoring program was presented and discussed. It was decided to hold off sending the results until the second round of sampling was complete. HDR will change the letter to incorporate the results of both sampling events as well as bolding the residence's results and highlighting the statement that the overall results are very low.

Distribution System and Source Samples

 25 distribution samples and 8 source samples were collected and the results of about 2/3 of them have been received to date.



- The results show that there was no detectable lead in the source samples.
- Jeff is completing the second half of the distribution system and source sampling this week. These results are expected in mid-June and are the last samples that need to be taken for the first round of the initial monitoring program.

c. Results to EPA

- A draft letter and technical memo for Patrick Chan (EPA) was presented that summarizes the results of the first round of sampling that has been completed to date. The technical memo will be updated and reissued when the results of the second distribution system and source sampling are received. There were no comments on the draft letter so HDR will issue the letter and technical memo to EPA as soon as the last four tap sample results are received.

II. SECOND ROUND OF INITIAL MONITORING PROGRAM

a. Scheduling

- The tap water sampling for the second round is scheduled to start on June 21 and the majority of the samples should be collected by June 24.
- The first half of the distribution system and source water sampling should also occur during the week of June 21.
- The second half of the distribution system and source water sampling should occur three weeks later during the weeks of either July 12 or July 19.

b. Resident Notification

 HDR will prepare a letter to the 130 homes sampled during the first round which notifies them of the dates for the second round of sampling. After the City's review and preparation of an original letter, HDR will make copies and send the letters.

III. MATERIALS SURVEY AND SAMPLING PLAN

a. Materials Survey

 HDR presented a draft technical memo for the City's review which summarizes the activities of the materials survey and how the tap water sample sites were located. The technical memo is missing a letter from the City's distribution department (to the production department) that states that no known lead lines exist, lead lines are replaced as they are found and the neighborhood is checked for additional lead lines, and when was the last time lead lines were found (if someone can recall). Bill B. said he would make sure the letter was written.

b. Sampling Plan

 HDR presented a draft technical memo for the City's review which summarizes the procedures used to collect and analyze the tap water, distribution system, and source water samples.

IV. GENERAL ITEMS

- a. Report Presentation Format
 - Perri S. suggested that HDR prepare a binder in which all reports, sampling results, and EPA correspondence would be contained. The binder would have sections for the following items:

Materials Survey
Sampling Plans
First Round Initial Monitoring Results
Second Round Initial Monitoring Results
Follow-up Monitoring Results
EPA/DHS Correspondence

 Three copies of this binder will be prepared for the City along with one copy each for EPA and DHS.



Mr. Patrick Chan U.S. Environmental Protection Agency 75 Hawthorne Street San Francisco, California 94105-3901

Subject:

City of Fresno - Lead and Copper Rule, Preliminary Report of Second Round

Monitoring Results

Dear Patrick:

The intent of this letter is to transmit to you a preliminary summary of the tap water sample analysis results and the first set of water quality analysis results. A final report detailing these results will follow after the second round of water quality analysis is complete. The second round of samples were collected between July 20 and 22, and the results are expected near the end of August.

The City of Fresno sampled 130 of the 131 residences in their Tier 1-C sample pool for lead and copper. The one sample not collected was from a vacant residence. In addition, water quality samples were collected from 25 distribution system sites and 7 pump stations that were identified as the source water for the residences sampled. The attached technical memorandum summarizes all of these results.

If you have any questions concerning these results, please call me at 916-939-4100.

Very truly yours,

Perri Standish-Lee

Y. Standish Lee

Vice President

PPSL/jh

cc: Martin McIntyre - City of Fresno

Doug Kirk - City of Fresno
Bill Dunn - City of Fresno
Matt Bullis - City of Fresno

Cindy Forbes - California Department of Health Services

Steve Reiber - HDR Engineering, Inc. Dave Murbach - HDR Engineering, Inc.

CITY OF FRESNO, WATER DIVISION LEAD AND COPPER RULE COMPLIANCE MONITORING RESULTS OF INITIAL MONITORING - SECOND ROUND TECHNICAL MEMORANDUM

TAP WATER SAMPLE RESULTS

Table 1 presents the results of the tap water analysis for lead and copper. The table lists the lead and copper concentrations in descending order. This was done in order to determine the 90th percentile levels as required by the Lead and Copper Rule.

Lead Results

The 90th percentile lead level was determined by multiplying the number of samples taken by $0.9 (130 \times 0.9 = 117)$. The 90th percentile lead level for the City of Fresno samples is 0.008 mg/L which is below the EPA action level of 0.015 mg/L.

Copper Results

The 90th percentile copper level was determined in the same way as for lead. The 90th percentile copper level for the City of Fresno is 0.43 mg/L which is below the EPA action level of 1.3 mg/L.

DISTRIBUTION SYSTEM AND SOURCE SAMPLE RESULTS

Water quality analysis was performed on 25 distribution system locations and 7 points of entry to the distribution system. The results of this analysis are summarized in Tables 2 and 3. These results include both sets of sampling that were performed in conjunction with the first round of initial monitoring.

The lead concentration of the source water was less than 0.001 mg/L at 6 of the 7 sample locations. At W-KVDS, lead was detected in the water quality samples and reported at 0.0025 mg/L. It should be noted that the laboratory reports lead levels between 0.001 and 0.0049 mg/L as 0.0025 mg/L.

TABLE 1. TAP WATER SAMPLE ANALYSIS

Ranking		Sample No.	Lead mg/L		Sample No.	Copper mg/L
130		161	0.110		153	0.76
129		155	0.072		27	0.62
128		151	0.053		80	0.57
127		148	0.046		130	0.56
126		173	0.025		139	0.56
125		142	0.021		137	0.55
124		87	0.012		166	0.55
123		145	0.012		10	0.54
122		170	0.012		142	0.51
121		114	0.011		83	0.49
120		18	0.010		175	0.49
119		92	0.010		141	0.48
118		154	0.009		152	0.46
117	90th %	25	0.008	90th %	18	0.43
116		119	0.008		128	0.43
115		140	0.008		143	0.43
114		65	0.007		46	0.42
113		120	0.007		113	0.42
112		79	0.006		172	0.42
111		102	0.006		173	0.42
110		135	0.006		146	0.41
109		30	0.005		149	0.40
108		108	0.005		55	0.39
107		1	0.0025*		114	0.39
106		2	0.0025*		169	0.39
105		10	0.0025*		127	0.38
104		11	0.0025*		81	0.37
103		20	0.0025*		129	0.37
102		26	0.0025*		147	0.36
101		27	0.0025*		159	0.36
100		38	0.0025*		79	0.35
99		39	0.0025*		51	0.34
98		41	0.0025*		63	0.34
97		43	0.0025*		88	0.34
96		48	0.0025*		89	0.34
95		53	0.0025*		90	0.32
94		54	0.0025*		102	0.31
93		60	0.0025*		163	0.31
92		64	0.0025*		171	0.31
91		66	0.0025*		1	0.29
90		75	0.0025*		107	0.29
89		78	0.0025*		161	0.29
88		81	0.0025*		168	0.29
87		89	0.0025*		52	0.28
86		90	0.0025*		53	0.28
85		93	0.0025*		111	0.28
84		100	0.0025*	1	132	0.28

TABLE 1. TAP WATER SAMPLE ANALYSIS

Ranking	Sample No.	Lead mg/L	Sample No.	Copper mg/L
83	109	0.0025*	42	0.27
82	112	0.0025*	64	0.27
81	116	0.0025*	131	0.27
80	117	0.0025*	140	0.27
79	118	0.0025*	150	0.27
78	122	0.0025*	54	0.26
77	127	0.0025*	66	0.26
		0.0025*	174	
76	128	0.0025*		0.26
75 74	130	0.0025*	25 109	0.25
73	131	0.0025*	112	0.25
72	132	0.0025*	125	0.25
71				
	134	0.0025*	100	0.24
70	137	0.0025*	120	0.24
69	139	0.0025*	133	0.24
68	141	0.0025*	155	0.24
67	143	0.0025*	2	0.23
66	146	0.0025*	76	0.23
65	147	0.0025*	119	0.23
64	149	0.0025*	30	0.22
63	150	0.0025*	48	0.22
62	152	0.0025*	65	0.22
61	156	0.0025*	40	0.21
60	163	0.0025*	50	0.21
59	166	0,0025*	105	0.21
58	167	0.0025*	124	0.21
57	168	0.0025*	39	0.20
56	171	0.0025*	58	0.20
55	172	0.0025*	78	0.20
54	175	0.0025*	98	0.20
53	176	0.0025*	60	0.19
52	3	<0.001	71	0.19
51	4	<0.001	115	0.19
50	21	<0.001	134	0.19
49	36	<0.001	144	0.19
48	40	<0.001	151	0.19
47	42	<0.001	156	0.19
46	46	< 0.001	61	0.18
45	50	< 0.001	93	0.18
44	51	< 0.001	136	0.18
43	52	< 0.001	148	0.18
42	55	< 0.001	38	0.17
41	56	< 0.001	41	0.17
40	57	< 0.001	69	0.17
39	58	< 0.001	110	0.17
38	61	<0.001	126	0.17
37	63	< 0.001	158	0.17

TABLE 1. TAP WATER SAMPLE ANALYSIS

Ranking	Sample No.	Lead mg/L	Sample No.	Copper mg/L
36	67	<0.001	170	0.17
35	69	<0.001	4	0.16
34	71	<0.001	26	0.16
33	72	<0.001	87	0.16
32	74	<0.001	108	0.16
31	76	<0.001	91	0.15
30	80	<0.001	92	0.15
29	83	<0.001	135	0.15
28	84	<0.001	36	0.14
27	86	<0.001	56	0.14
26	88	<0.001	57	0.14
25	91	<0.001	86	0.13
24	95	<0.001	118	0.13
23	98	<0.001	122	0.13
22	103	<0.001	145	0.13
21	105	< 0.001	164	0.13
20	107	< 0.001	67	0.12
19	110	< 0.001	95	0.12
18	111	<0.001	176	0.12
17	113	<0.001	103	0.10
16	115	< 0.001	84	0.10
15	123	< 0.001	72	0.10
14	124	< 0.001	11	0.09
13	125	< 0.001	43	0.09
12	126	<0.001	75	0.08
11	129	< 0.001	117	0.07
10	136	< 0.001	123	0.06
9	144	< 0.001	167	0.06
8	153	< 0.001	165	0.05
7	157	< 0.001	154	0.025**
6	158	< 0.001	116	0.025**
5	159	< 0.001	74	0.025**
4	164	< 0.001	21	0.025**
3	165	< 0.001	20	0.025**
2	169	< 0.001	3	0.025**
1	174	< 0.001	157	< 0.01

^{*} The laboratory reports lead values between 0.0010 and 0.0049 mg/L as 0.0025 mg/L.

^{**} The laboratory reports copper values between 0.010 and 0.049 mg/L as 0.025 mg/L.

TABLE 2. DISTRIBUTION SYSTEM WATER QUALITY ANALYSIS

		Samples Take	n 6/23/93 -	7/1/93	
Sample No.	Alkalinity (mg/L CaCO3)	Calcium (mg/L)	pН	Temp. (deg C)	Cond. (mS/cm)
D-31	140	29	7.26	26.1	0.35
D-34	120	29	7.68	28.6	0.30
D-36	110	26	7.51	25.8	0.30
D-66	75	16	7.47	25.4	0.21
D-68	170	44	7.13	24.9	0.47
D-69	160	38	7.03	26.7	0.47
D-70	120	28	7.12	29.7	0.28
D-76	91	21	7.30	24.5	0.23
D-79	130	28	7.43	26.5	0.27
D-81	110	26	7.24	29.7	0.29
D-82	110	26	7.34	25.6	0.28
D-116	110	29	7.03	25.8	0.36
D-117	130	29	7.32	28.1	0.31
D-126	99	24	7.74	23.9	0.29
D-128	110	21	7.21	26.2	0.27
D-129	120	27	7.25	25.8	0.28
D-130	110	24	7.28	25.2	0.26
D-133	99	22	7.45	25.4	0.25
D-134	93	21	7.50	26.7	0.23
D-155	110	24	7.23	29.7	0.25
D-156	110	21	7.20	29.5	0.28
D-158	120	30	7.16	27.8	0.35
D-159	130	29	7.22	25.7	0.37
D-181	120	27	7.32	26.6	0.31
D-202	77	20	7.68	22.8	0.23
Automore	115	26	7.22	2/5	0.20

Average	115	26	7.32	26.5	0.30
Minimum	75	16	7.03	22.8	0.21
Maximum	170	44	7.74	29.7	0.47

TABLE 3. SOURCE WATER QUALITY ANALYSIS

			Samples Ta	ken 6/23/93 - *	7/1/93		
Sample No.	Lead (mg/L)	Copper (mg/L)	Alkalinity (mg/L CaCO3)	Calcium (mg/L)	pН	Temp. (deg C)	Cond. (mS/cm)
W-79	< 0.001	< 0.01	120	27	7.43	26.5	0.27
W-86	NS	NS	NS	NS	NS	NS	NS
W-91	< 0.001	< 0.01	97	23	7.63	22.3	0.27
W-97	< 0.001	< 0.01	120	27	7.36	22.9	0.30
W-99	< 0.001	0.025*	85	18	7.53	22.1	0.27
W-131	< 0.001	< 0.01	95	21	7,42	24.4	0.26
W-140	< 0.001	< 0.01	110	31	7.12	23.4	0.36
W-KVDS	0.0025**	0.025*	110	26	7.46	23,3	0.28
Average	0.0012	0.014	105	25	7.42	23.6	0.29
Minimum	<0.001	< 0.01	85	18	7.12	22.1	0.26
Maximum	0.0025**	0.025*	120	31	7.63	26.5	0.36

NS = No sample taken because pump was out of service. With this pump out of service, W-131 is the source for the area served by W-86.

^{*} The laboratory reports copper values between 0.010 and 0.049 mg/L as 0.025 mg/L.

^{**} The laboratory reports lead values between 0.0010 and 0.0049 mg/L as 0.0025 mg/L.

TABLE 5. Distribution System Sample Locations

Distribution System Sample Number	Distribution System Location Number	Address	Associated Pump Station Number
7/2/1	79 (b)	(6)	91
7/2/2	129		79, 97
7/20/853	- 126		KV-DS
7/2/4	68		86, 131
7/215	70		86
7/216	116		140
7/20 7	133		99
7/21 8	181		97
7/2/9	31 29		General* (91)
7/2010	130		General* (91)
7/2011	-366		General* (KV-DS)
7/2/12	- 159		General* (86, 140)
13 د ٦	158		General* (140)
7/2014	134		General* (99)
7/2015	. 34 547		General* (99)
7/2016	, 36		General* (99)
7/2/17 -	202		General* (97)
7/2018	81		General* (97)
7/2/19	155		General* (131)
7/2120	69		General* (131)
7/3021	82		· General**
7/2122	156		General**
7/2/23	128		General**
7/2/24	117		General**
9/20 25	. 76		General**

^{*} General distribution system sample location, number in () represents remotely associated pump station.

^{**} General distribution system sample located in the same region as the tap sample locations.

TABLE 3. Source Water Sample Locations

Pump Station Number	Address	Location	Associated Tap Sample Numbers
91	(b) (9)		1,3
79			2
KV-DS	-		4,144,158
86	-		10
140	-		Tract B*
99			Tract C*
97	_		Tract D*
131			Tract E*

^{*} Tract B contains 63 sample sites, Tract C contains 46 sample sites, Tract D contains 29 sample sites, and Tract E contains 17 sample sites.

TABLE 4. Source Sample Analytical Methods

Compound	EPA Method	Units	DLR*	Sample Container
Copper	200.7	mg/L	0.010	1 L polyethylene
Lead	239.2	mg/L	0.001	taken with copper
Calcium	200.7	mg/L	0.1	200 mL polyethylene
Alkalinity	310.1	mg/L as CaCO ₃	1.0	100 mL polyethylene

* Detection limit for the purposes of reporting

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: W-140 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _ Q. M

Date = $\frac{9/20/93}{100}$ Time = $\frac{9.20}{100}$ AM

pH = 7.18

Temp. = 22.7 deg C

Conductivity = 0,31 mS/cm

LOCATION: $\sqrt{-9}$ (W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = £. B.

Date = 7/20/9 Time = 10:10 A

pH = 7.50

Temp. = 22.4 deg C

Conductivity = 0.26 ms/cm

LOCATION: D-76 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = F. B

Date = $\frac{7/20/93}{7.53}$ Time = $\frac{/2:35p}{7.53}$

Temp. = 27.1 deg C

Conductivity = 0.3 / mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-126 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 4.71

pH = 7.39

Temp. = 27.3 deg C

Conductivity = 0.26 mS/cm

LOCATION: W- (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 22.

Date = $\frac{7/26^{\frac{1}{2}}2}{2}$ Time = $\frac{11.56}{2}$

pH = 7.37

Temp. = $2^{4} \cdot 3$ deg C

Conductivity = 0,27 mS/cm

LOCATION: D-66 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Date =
$$\frac{7/26/9}{3}$$
 Time = $\frac{1.20}{20}$ PIN

pH = 7:37

Temp. = 27.4 deg C

Conductivity = $f_1, 23$ mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: - /30 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 5.3.

Date = 7/20/93 Time = 1:302

pH= 7.33

Temp. = 24.7 deg C

Conductivity = 0,27 mS/cm

LOCATION: (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = A. M.

Date = $\frac{7}{20.63}$ Time = $\frac{1.450}{1.450}$

pH = 7.50

Temp. = \$5.7 deg C

Conductivity = 0, 27 mS/cm

LOCATION: (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials =

Date = 2002 Time = 1:55p

pH = 7.45

Temp. = 25.1 deg C

Conductivity = 0.30 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-34 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = 9.1

Date = $\frac{7/20/9}{3}$ Time = $\frac{2:15}{15}$ Pill

pH = 7.67

Temp. = 32.0 deg C

Conductivity = 0.27 mS/cm

LOCATION: D-36 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = F.B.

Date = $\frac{7}{20}$ Time = $\frac{21300}{2}$

pH = 7.60

Temp. = 26.5 deg C

Conductivity = 2,28 mS/cm

LOCATION: (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials =

Date = 1 20 1 2 Time = < 145 p

pH = 7.69

Temp. = 27.7 deg C

Conductivity = 2, 26 mS/cm

LOCATION: D-134	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	<u> </u>
Date =	$\frac{7/20193}{}$ Time = $\frac{3/95p}{}$
pH =	7.53
Temp. =	<u>29.7</u> deg C
Conductivity =	<u>D.⊋3</u> mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
рН =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: W-7 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _ A. M.

Date = 1 8 10 4

PH = 7.62

Temp. = 22.7 deg C

Conductivity = 0.24 mS/cm

LOCATION: 0- 202 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = $\underline{\mathcal{E}, \mathcal{S}}$

Date = 7 2193 Time = 2:40A

pH = (, 41 -)

Temp. = A 4. A deg C

Conductivity = 5, 27 mS/cm

LOCATION: (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials =

Date = 7 21 93 Time = 8:50 A

pH = ______

Temp. = 2 1 dea C

Conductivity = 0.8% mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-68 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = AM

Date = $\frac{7/21/9}{9}$ 3 Time = $\frac{10}{10}$ AM

pH = 7.21

Temp. = 22,5 deg C

Conductivity = 0.41 mS/cm

LOCATION: (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = $\overline{\mathcal{E}}$.

Date = 7/21/43 Time = 7/21/04

pH = 7.17

Temp. = 27.0 deg C

Conductivity = 0.37 mS/cm

LOCATION: <u>W-131</u> (W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = $\mathcal{A} \cdot \mathcal{N}$.

Date = $\frac{7/21/9}{3}$ Time = $\frac{10.30}{4}$ PW1

pH = 7.35

Temp. = 24.0 deg C

Conductivity = 0.23 mS/cm

LOCATION: 155	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	
pH =	7,45
Temp. =	<u>⊇ 7, </u>
Conductivity =	<u>0.21</u> mS/cm
LOCATION: 55 117	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	B.17
Date =	7/21/93 Time = $10:50$ AM
pH =	7,30
Temp. =	27.7 deg C
Conductivity =	0.26 mS/cm
LOCATION: 1-00	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	\$, B
Date =	7)21 1/13 Time = A
	7.24
	<u>≈5.</u> deg C
Conductivity =	<u>0.27</u> mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D-159 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Date = $\frac{7/31/9}{3}$ Time = 11:15 AM

pH = 7,20

Temp. = 26.5 deg C

Conductivity = 0.29 mS/cm

LOCATION: 0-158 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = FJ

Date = 7/21/75 Time = 11:25A

pH = 7.16

Temp. = 27,0 deg C

Conductivity = 0.35 mS/cm

LOCATION: D-116 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = \mathcal{G} . \mathcal{M} .

Date = 7/21/93 Time = 11:40 AM

 $pH = _{7.11}$

Temp. = 26.7 deg C

Conductivity = 0.34 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: D- 156 (W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _E, 5.

Date =
$$\frac{7.2193}{200}$$
 Time = $\frac{1.935}{200}$

$$pH = 7.17$$

Temp. = $\frac{28.1}{\text{deg C}}$

Conductivity = 0.35 ms/cm

LOCATION: (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = <u>Q. N.</u>

Date = $\frac{7/21/93}{2}$ Time = $\frac{1!25}{9}$

pH = 7,13

Temp. = 25, 4 deg C

Conductivity = 0.32 mS/cm

LOCATION: (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials =

$$pH = 7.20$$

Temp. = 25.1 deg C

Conductivity = 0.32 mS/cm

DISTRIBUTION & WELL SAMPLE FIELD LOG

LOCATION: 0-79 (W-## = WELL SITE)

(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = _ A. M.

Date = $\frac{7/21/9}{3}$ Time = $\frac{1:50}{9}$

pH = 7.43

Temp. = 23.6 deg C

Conductivity = 0,26 mS/cm

LOCATION: W - 7 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Date = 7/21/93 Time = 27:07 5

pH = 7.25

Temp. = 23.3 deg C

Conductivity = 0.29 mS/cm

LOCATION: D-181 (W-## = WELL SITE)
(D-## = DISTRIBUTION SYSTEM SITE)

Samplers Initials = \mathcal{A} . \mathcal{N} .

Date = $\frac{7/21/93}{2}$ Time = $\frac{2115}{9}$

DH = 7.32

Temp. = 28.4 deg C

Conductivity = 10.29 mS/cm

LOCATION: D-129	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	<u>F.B.</u>
Date =	7/2:193 Time = 2:25p
pH =	7.26
Temp. =	31,4 deg C
Conductivity =	<u>0,3</u> ms/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

LOCATION: W-86	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	E. B.
Date =	7 88 1 1 Time = 9.05A
pH =	7.41
Temp. =	<u> </u>
Conductivity =	0.29 ms/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm
LOCATION:	(W-## = WELL SITE) (D-## = DISTRIBUTION SYSTEM SITE)
Samplers Initials =	
Date =	Time =
pH =	
Temp. =	deg C
Conductivity =	mS/cm

La Re	Received by AC	100	er		Date: 7 / 20/ 53 Time: 15: 40 am 1000	: 40 am pan No of Samples: 26 23
Sar	Sample Detail					
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00	7/20/93	1:45	3		7-81	
9	7/20/93	1:55 PM M	N		D-82	< × ×
0	10 7/20/93 2:15 PM M	2:15 PM	3		D-34	< >

Comments or Instructions:

Sample Seals: Containers:

TYPE: S

Sample Activity Form

City of Fresno - Water Division, 1910 E. University Ave., Fresno, CA 93703

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D-134		3	3:05 PM	7/20/93	13
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Date: 7/20/93 Time: 15:50 am 6m		pro	な、ど	Received by: 2	Rec
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BSR Log #: (9: Sample Seals: Containers:

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TYPE: S Due Date:

No of Samples:	Lao Reference:	acotatory.
	I St Brewning 16 C	Taboratory BSK
30 am pm	Date: 712193 Time: 15:30 am pm	Received by: J. a. Vera
Sampled by: HEThus NAVARRO Relinquished by: Wilher Thuring Date: 7/21/93 Time: 15: 30 am p	Relinquished by: Lathur Hurr	ampled by: HETTINE NAVAERO
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Sample Detail

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Comments or Instructions:

Sample Seals: Containers:
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Sampled by: ARThUR MAJARRO	Relinquished by: Listher Muzens Date: 712/1	2/1 43Time: 15:30 um pn
Received by: J. C. Verr	Date: 7/20/23 Time: 15: 30 am pm	
aboratory: BSK		No of Samples:
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12 7/21/53 11:40 M M	D-116	XX
13 7/21/93 1:00 P M -:	D-156	×××
14 7/21/93 1:25 PM	D-128	X X
15 7/21/93 1:35 P M	D-31	×
16 7/21/93 1:50P M	D-79	X
17 7/21/93 2:00 P.S.	w-79	XXXX
18 7/21/93 Z:ISP M	D-181	XX
7/21/93	D-129	××
30 7/20/93 10:30P AD	78	×××

Comments or Instructions:

Sample Seals:

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